## Mark schemes

0	1	
w		-

(a) **Level 3**: The design/plan would lead to the production of a valid outcome. All key steps are identified and logically sequenced.

5-6

**Level 2:** The design/plan would not necessarily lead to a valid outcome. Most steps are identified, but the plan is not fully logically sequenced.

3-4

**Level 1**: The design/plan would not lead to a valid outcome. Some relevant steps are identified, but links are not made clear.

1-2

## No relevant content

### **Indicative content**

#### Method

- draw (pencil) start line on (chromatography) paper
- place spot of food colouring on start line
- use of suitable solvent
- place solvent in beaker / container
- place (chromatography) paper in beaker / container
- so (chromatography) paper is in solvent
- but solvent is below start line
- use a lid
- wait for solvent to travel up the (chromatography) paper (until near top)
- mark solvent front
- dry the (chromatography) paper

# Measurements

- measure distance between start line and centre of spot
- measure distance between start line and solvent front
- use of measurements to determine R<sub>f</sub> value
- (b) different solvent used

1

(c) paper

[8]

Q2.

(a) <sup>2.7</sup>/<sub>9.0</sub>

1

= 0.3

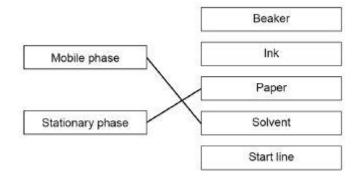
1

ignore units

(b) **C** and **D** 

1

(c)



additional line from a box on the left negates the mark for that box

1

1

(d) 17:3

1

(e) formulation

1

(f) the R<sub>f</sub> value would stay the same

[8]

Q3.

1

(distance moved by solvent =)  $\frac{5.7}{0.60}$ 

1

= 9.5 (cm)

1

(b) some of the compounds are colourless (in solution)

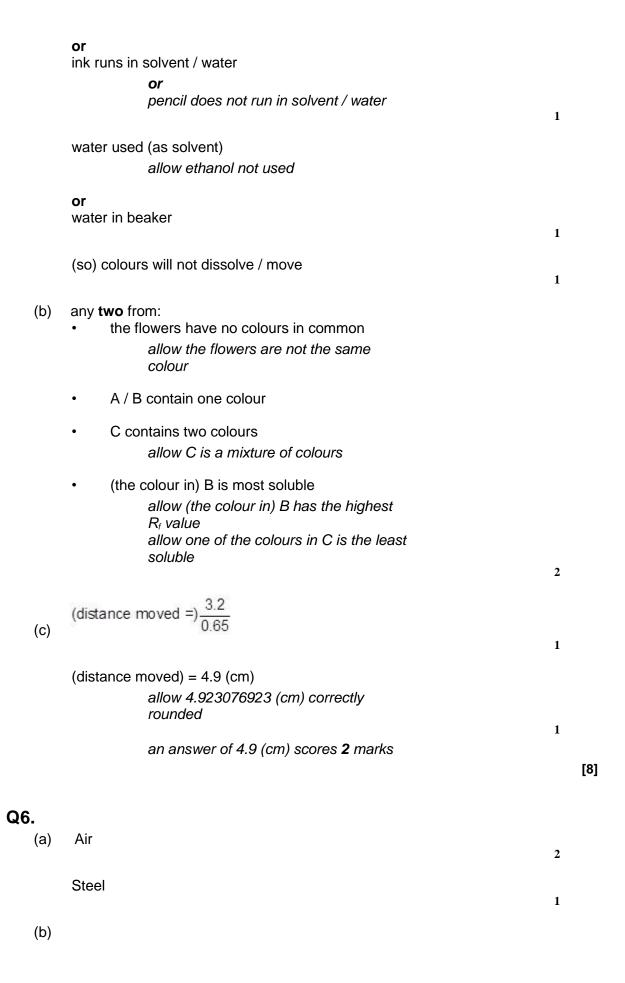
or

dyes / compounds have the same R<sub>f</sub> values

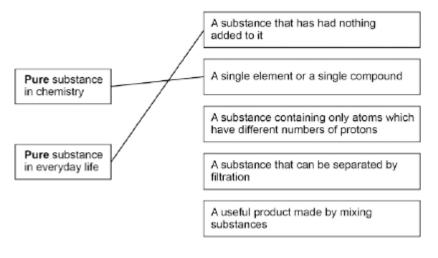
allow there are only two compounds
that are coloured (in solution)

1

(c)	allow the solvent front to travel further	1	
	use a different solvent	1	
(d)	so that the (shade of) green is the same  allow because the green ink is a formulation	1	
(e)	the dye is less soluble in the new solvent and more attracted to the new paper	1	[8]
Q4.			
(a)	crush the flower	1	
	use more flowers	1	
(b)	the start line is drawn in ink	1	
	uses water as the solvent	1	
(c)	flower A contains a single pure colour	1	
	the colour in flower C is a mixture	1	
(d)	7.2 9.0	1	
	= 0.8		
	an answer of 0.8 scores <b>2</b> marks ignore units	1	
			[8]
<b>Q5.</b> (a)	start line drawn in ink  allow start line should have been drawn in pencil	1	
	(so) ink dissolves (as) pencil does not dissolve		



1



Allow 1 mark for the correct meanings linked to context but incorrect way around

(c) Damp litmus paper turns white

(d) Iron(III)

1
[6]

**Q7**.

(a) The start line was drawn in ink

The water level was above the spots

(b) 3

(c) A

(d) (distance moved by dye A) 38 (mm)

allow values in range 36-40

1

102 (mm)

allow values in range 101-103

38

allow ecf from Table 1

0.37254 ...

allow values in range 0.35 - 0.39

(distance from start line to solvent front)

102

		1	
	0.37  accept 0.37 with no working shown for 5 marks	1	[9]
<b>Q8.</b> (a)	water level above the start line and start line drawn in ink allow water level too high	1	
	water level food colours would dissolve into water or start line the ink would 'run' on the paper		
(b)	(distance moved by <b>A</b> ) 2.8cm <b>and</b> 8.2 cm (distance moved by solvent) allow values in range 2.7 - 2.9 cm and 8.1 - 8.3 cm	1	
	2.8 8.2	1	
	allow 0.33 or 0.35  allow ecf from incorrect measurement to final answer for <b>2</b> marks if given to 2 significant figures accept 0.34 without working shown for <b>3</b> marks	1	
(c)	6.6 cm  allow values between 6.48 and 6.64 cm	1	
(d)	solvent moves through paper	1	
	different dyes have different solubilities in solvent	1	
	and different attractions for the paper	1	
	and so are carried different distances	1	
(e)	calcium ions  allow Ca <sup>2+</sup>		

		1
	sodium ions	
	allow Na+	1
		1
(f)	two different colours	
	or Ca <sup>2+</sup> / one is orange-red and Na+ / the other is yellow	
	allow brick red for Ca <sup>2+</sup> and / or orange for Na+	
	allow incorrect colours if consistent with answer to	
	7.5	1
	(so) colours mix or	
	(so) one colour masks the other	
		1
(g)	(Student A was incorrect)	
	because sodium compounds are white not green or	
	because sodium carbonate is soluble	
		1
	so can't contain sodium ions	
		1
	(Student <b>B</b> was incorrect)	
	because adding acid to carbonate produces carbon dioxide	1
	so must contain carbonate not chloride ions	1
		[18]